



LABORATORY CHEMICALS AND CONSUMABLES

MATERIAL SAFETY DATA SHEET

PENTANE

1. Chemical Product and Company information.

Product name: Pentane

Contact Information:

Radchem cc
PO Box 166982
Brackendowns
Alberton 1454
Telephone : **011 867 3726 / 2864**

2. Hazard Identification

Hazardous in case of skin contact (irritant), of eye contact (irritant), of ingestion, of inhalation. Slightly hazardous in case of skin contact (permeator).

3. Composition / information on ingredients

CAS #: 109-66-0

Synonym: Amyl Hydride

Chemical Name: Pentane

Chemical Formula: C₅-H₁₂

4. First Aid Measures

Eye Contact: Check for and remove any contact lenses. In case of contact, immediately flush eyes with plenty of water for at least 15 minutes. Cold water may be used. WARM water MUST be used. Get medical attention.

Skin Contact: In case of contact, immediately flush skin with plenty of water. Cover the irritated skin with an emollient. Remove contaminated clothing and shoes. Wash clothing before reuse. Thoroughly clean shoes before reuse. Get medical attention.

Serious Skin Contact: Wash with a disinfectant soap and cover the contaminated skin with an anti-bacterial cream. Seek medical attention.

Inhalation: If inhaled, remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Get medical attention if symptoms appear.



Serious Inhalation: Evacuate the victim to a safe area as soon as possible. Loosen tight clothing such as a collar, tie, belt or waistband. If breathing is difficult, administer oxygen. If the victim is not breathing, perform mouth-to-mouth resuscitation. Seek medical attention.

Ingestion: If swallowed, do NOT induce vomiting. Never give anything by mouth to an unconscious person. Aspiration hazard if swallowed- can enter lungs and cause damage. Loosen tight clothing such as a collar, tie, belt or waistband. Get medical attention.

Serious Ingestion: Not available.

5. Fire-fighting measures

Flammability of the Product: Flammable

Fire Hazards in Presence of Various Substances: Extremely flammable in presence of open flames and sparks, of heat. Flammable in presence of oxidizing materials. Non flammable in presence of shocks.

Explosion Hazards in Presence of Various Substances: Risks of explosion of the product in presence of static discharge: Not available. Slightly explosive in presence of heat, of oxidizing materials. Non-explosive in presence of shocks.

Fire Fighting Media and Instructions: Flammable liquid, soluble or dispersed in water. SMALL FIRE: Use DRY chemical powder. LARGE FIRE: Use alcohol foam, water spray or fog. Cool containing vessels with water jet in order to prevent pressure build-up, auto ignition or explosion.

Special Remarks on Fire Hazards: EXTREMELY FLAMMABLE. Vapour may travel considerable distance to source of ignition and flash back.

Special Remarks on Explosion Hazards: Not applicable

6. Accidental release measures

Small Spill: Absorb with an inert material and put the spilled material in an appropriate waste disposal.

Large Spill: Flammable liquid. Keep away from heat. Keep away from sources of ignition. Stop leak if without risk. Absorb with DRY earth, sand or other non-combustible material. Do not touch spilled material. Prevent entry into sewers, basements or confined areas; dike if needed. Be careful that the product is not present at a concentration level above TLV. Check TLV on the MSDS and with local authorities.

7. Handling and storage

Precautions: Keep away from heat. Keep away from sources of ignition. Ground all equipment containing material. Do not ingest. Do not breathe gas/fumes/ vapour/spray. Wear suitable protective clothing. In case of insufficient ventilation, wear suitable respiratory equipment. If ingested, seek medical advice immediately and show the container or the label. Avoid contact with skin and eyes. Keep away from incompatibles such as oxidizing agents.

Storage: Store in a segregated and approved area. Keep container in a cool, well-ventilated area. Keep container tightly closed and sealed until ready for use. Avoid all possible sources of ignition (spark or flame). Do not store above 25°C

8. Exposure controls/personal protection

Engineering Controls: Provide exhaust ventilation or other engineering controls to keep the airborne concentrations of vapours below their respective threshold limit value. Ensure that eyewash stations and safety showers are proximal to the work-station location.

Personal Protection: Splash goggles. Lab coat. Vapour respirator. Be sure to use an approved/certified respirator or equivalent. Gloves.



Personal Protection in Case of a Large Spill: Splash goggles. Full suit. Vapour respirator. Boots. Gloves. A self contained breathing apparatus should be used to avoid inhalation of the product. Suggested protective clothing might not be sufficient; consult a specialist BEFORE handling this product.

9. Physical and chemical properties

Physical state and appearance: Liquid

Odour: Pleasant. Gasoline-like

Taste: Not available

Colour: Colourless

Boiling Point: 36.1°C

Melting Point: -130°C

Critical Temperature: Not available

Specific Gravity: 0.6262 (Water = 1)

Vapour Density: 2.49 (Air = 1)

Volatility: Not available

Odour Threshold: 2.2 ppm

Ionicity (in Water): Not available.

Dispersion Properties: See solubility in water, diethyl ether, acetone

Solubility: Partially soluble in diethyl ether, acetone. Very slightly soluble in cold water. Solubility in water: 0.36g/1 water @ 16 deg. C. Solubility in water: 9.9 g in 100 kg water @ 25 deg. C. Solubility in water: 0.04 g in 100 g of water @ 20 deg. C. Soluble in chloroform. Solubility in acetone, benzene, ethanol > 10%

10. Stability and reactivity

Stability: The product is stable.

Instability Temperature: Not available.

Conditions of Instability: Heat, ignition sources, incompatible materials.

Incompatibility with various substances: Reactive with oxidizing agents

Corrosivity: Non-corrosive in presence of glass

Special Remarks on Reactivity: Not available

Special Remarks on Corrosivity: It will attack some forms of plastics, rubber, and coatings

Polymerization: Will not occur.

11. Toxicological information

Routes of Entry: Absorbed through skin. Eye contact. Inhalation. Ingestion

Toxicity to Animals: WARNING: THE LC50 VALUES HEREUNDER ARE ESTIMATED ON THE BASIS OF A 4-HOUR EXPOSURE. Acute toxicity of the vapour (LC50): 364000 mg/m³ 4 hours [Rat].

Chronic Effects on Humans: May cause damage to the following organs: kidneys, the nervous system, liver, skin, central nervous system (CNS).

Other Toxic Effects on Humans: Hazardous in case of skin contact (irritant), of ingestion, of inhalation. Slightly hazardous in case of skin contact (permeator).



Special Remarks on Toxicity to Animals: Lowest Published Lethal Dose: LCL [Mouse] - Route: Inhalation; Dose 325 gm/m³/2H

Special Remarks on Chronic Effects on Humans: Not available

Special Remarks on other Toxic Effects on Humans: Acute Potential Health Effects: Skin: It can cause skin irritation with itching, drying erythema, hyper pigmentation, hyperemia, dermatitis, burning sensations, followed by formation of blisters. It may be absorbed by the skin and cause systemic effects. Eyes: It is a strong eye irritant. Symptoms may include pain, corneal irritation, and nystagmus Inhalation: It is a mild respiratory tract (nose, throat, lungs) irritant causing coughing, wheezing, and/or shortness of breath. Inhalation exposure to an airborne concentration of 5,000 ppm for 10 minutes appears to have no ill effect in humans, while 90,000 to 120,000 ppm can affect behaviour/central nervous system and cause symptoms of central nervous system depression and narcosis. Symptoms of central nervous system depression include nausea, headache, weakness, dizziness, excitement, confusion, light-headedness, sleepiness, seizures, inability to concentrate, loss of coordination and judgement, coma, and death with exposure to large amounts. It may also affect the cardiovascular system (dysrhythmias), and metabolism (weight loss/loss of appetite/anorexia). Airborne concentration of approximately 130,000 ppm can be fatal by asphyxiation; therefore, there is not a wide margin of safety concentrations causing central nervous system effects and death. Ingestion: Pulmonary aspiration of even a small amount can produce acute lung injury, potentially fatal chemical pneumonitis, and haemorrhage. In extreme cases, respiratory arrest secondary to hypoxia following pneumonitis may occur. It may also affect behaviour/central nervous system (symptoms similar to acute inhalation), and cardiovascular system (symptoms similar to acute inhalation). Chronic Potential Health Effects: Skin: Prolonged or repeated skin contact can cause defatting dermatitis with dryness and cracking. Long-term dermal exposure may also cause kidney damage. Ingestion or inhalation: Prolonged or repeated ingestion or inhalation Can cause central nervous system damage. Central nervous system damage symptoms may include numbness, "pins and needles", and weakness of arms and legs. It may also cause liver damage, and kidney damage (renal tubular necrosis, glomerulonephritis, nephritis, proteinuria, hematuria) and may affect metabolism (weight loss).

12. Ecological information

Ecotoxicity: Not available

BOD5 and COD: Not available

Products of Biodegradation: Possibly hazardous short term degradation products are not likely. However, long term degradation products may arise

Toxicity of the Products of Biodegradation: The product itself and its products of degradation are not toxic

Special Remarks on the Products of Biodegradation: Not available

13. Disposal considerations

Waste Disposal: Waste must be disposed of in accordance with federal, state and local environmental control regulations.

14. Transport information

DOT Classification: CLASS 3: Flammable liquid

Identification: : Pentane UNNA: 1265 PG: II

Special Provisions for Transport: Not applicable

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